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Section 10.2 Mass and
the Mole Section 10.3

Moles of Compounds

Section 10.4 Empirical
and Molecular

Formulas Section 10.5

Formulas of Hydrates

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biological, physical.

Name the two

categories of science.

the science of different

kinds of matter and

how that matter can

change. Define

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study of carbon based

compounds. inorganic.

the study of non-

organic compounds.

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...

Elena is a chemist, a scientist who studies matter and the changes that it undergoes. Carl is very dependable. His teachers and his parents know that he is reliable and can be trusted. An air of jubilation surrounded the members of the science team as they

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received their medals
for first place in the
national competition.

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Physical & Chemical
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This video describes physical and chemical changes in matter.

Physical Changes.

Another way to think about this is that a physical change does not cause a substance to become a fundamentally different substance but a chemical change causes a substance to change into something chemically new.

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CHAPTER 10

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161 Section 10.1

Measuring Matter page

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320-324 Practice
Problems pages
323-324 1. Zinc (Zn) is
used to form a
corrosion-inhibiting
surface on galvanized
steel. Determine the
number of Zn atoms in
2.50 mol of Zn. 2.50
mol Zn

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8 Assessment
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Name 17 CHAPTER
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MASTE In the space at the left, write true if the statement is true; if the statement is false, change the italicized word or phrase to make it true 8 To determine the instantaneous rate, you must know the specific

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In this chapter you will learn- Concept of matter. Matter in our surroundings (full chapter). Matter in our surroundings
Notes. Notes and explanation. Short notes. Question and answer. Class 9 CBSE ...

**Matter In Our
Surroundings. Class
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Name _____ Date _____
CHAPTER
FOR Class Section 8.2
continued 9 What is
the relationship
between lattice energy
and the strength of the
attractive force holding
ions in place? a. The
more positive the
lattice energy is, the
greater the force. b.
The more negative the
lattice energy is, the
greater the force.

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Chemistry is central to
all of the sciences.
CHAPTER 1 Tartaric
Acid Crystals. MATTER
AND CHANGE 3
SECTION 1 OBJECTIVES
Define chemistry. List
examples of the
branches of chemistry.
Compare and contrast
basic research, applied

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research, and
technological
development.

CHAPTER 1 Matter and Change

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