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and Acknowledgments The first version of General Chemistry I Lab Manual was drafted by Ms. Khalida Jaffar and Dr. Khalid Omer. It was used in the academic year, 2012-2013.

General Chemistry CHM 1410L

Contents
1 Lab Manual Class 10 Science
1.1 Class 10 Science Lab Manual Features:
1.2 Experiments for Class 10 Science CBSE
1.3 Chemistry Practicals for Class 10
1.4 Physics Practicals for Class 10
1.5 Biology Practicals for Class 10
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approved a medical report to the General Chemistry Group Coordinator. 13.

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Alternatively 0.000003 m is also difficult to express without scientific notation. In order to convert this number to we move the decimal to behind the 3 (the first non-zero number) and add the power of 10 (here we moved the decimal 6 times so the exponent is -6) 3×10^{-6} . Here the exponent is negative because the number is very small (less than 1).

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Experimental Data and Results *Make sure you show all work and calculations to receive full credit. For each problem, find the ΔH for the target reaction below, given the following step reactions and subsequent ΔH values:

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Pre-Lab Name: Date: 1) Why are the titrations in this lab carried out in acidic solution? 2) If a student weighs out 0.465 g of oxalate unknown and it takes 26.8 ml. of permanganate solution to titrate it, how much oxalate unknown should the student weigh out in order to use 35.0 mL of permanganate solution?

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